The background of the slide is a collage of four images related to materials science and quantum simulations. The top-left image shows a complex, multi-colored molecular structure. The top-right image shows a network of green spheres connected by lines, representing a molecular or material structure. The bottom-left image shows a white, interconnected network of spheres and lines, possibly a crystal lattice or a complex molecule. The bottom-right image shows a similar white network with some yellow and blue highlights, suggesting a different material or simulation state.

# Quantum Simulations of Materials and Nanostructures

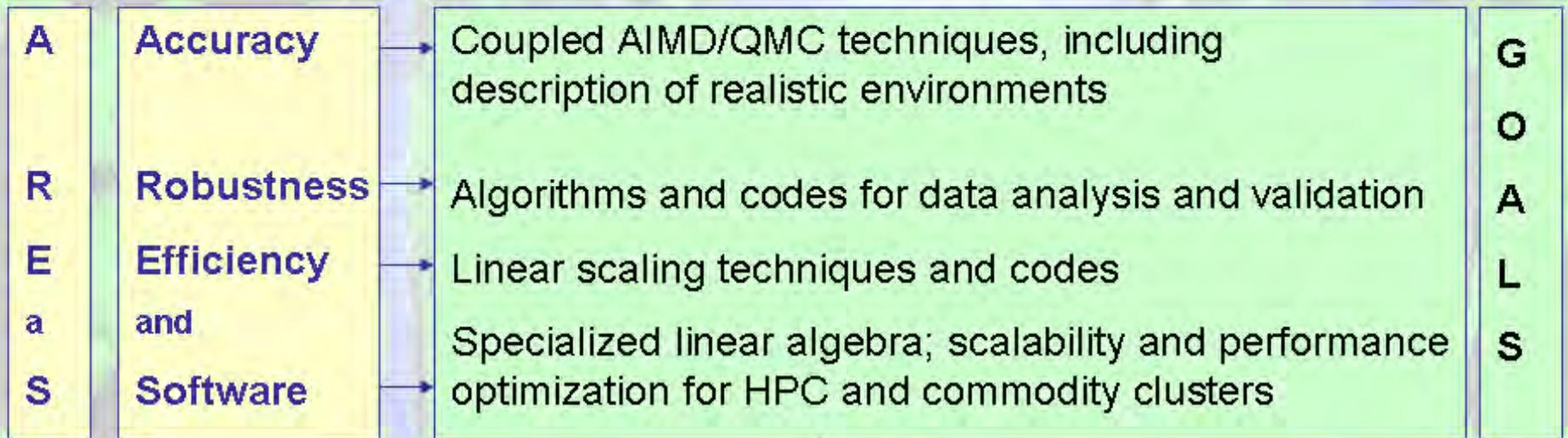
Giulia Galli

University of California, Davis

Collaborators: Z.Bai, F.Gygi and W.Pickett, UCD; W. Cai, Stanford U.; D.Ceperley, UIUC; N.Marzari, MIT; N.Spaldin, UCSB; E.Schwegler and J.-L.Fattebert, LLNL.

# Quantum Simulations of Materials and Nanostructures (Q-SIMAN)

*Predict and design molecular and materials properties with controllable accuracy, from first principles.*



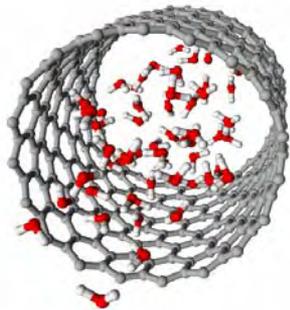
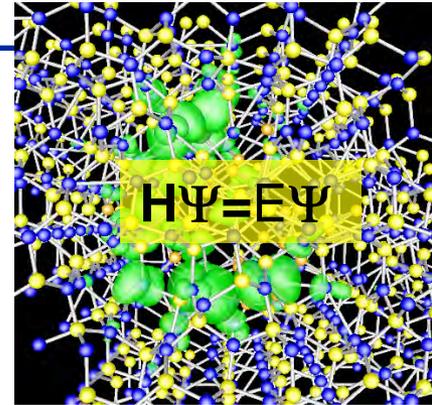
*Grand Challenge Applications to Materials and Nanostructures*





# Predictive modeling with controllable accuracy: a microscopic view in a nutshell

- We solve **approximate form of the Schrodinger equation** describing electron-electron and electron-ion interactions in molecules, condensed systems and nanostructures – **no input from experiment.**



- We couple our electronic structure description to **statistical mechanical techniques** to describe finite temperature properties and various thermodynamic conditions– **ab-initio Molecular Dynamics**; Monte Carlo

- We devise **strategies and algorithms** to compute **‘complex’ properties in realistic environments**; this encompasses:

**theoretical, algorithmic and software developments**





# Outline

---

- **Is computational materials science any good? Any promising progress?**
  - A couple of examples
    - **Validation**
    - **Impact**

- Our **strategy for predictive computations**
  - On different fronts, in parallel
- Some **grand challenge applications**



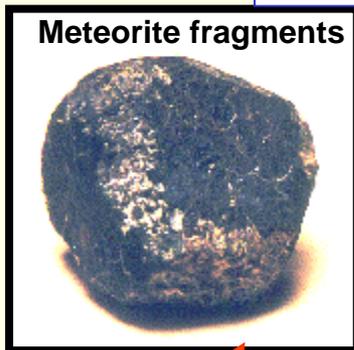
# Terrestrial and extra-terrestrial ultra-disperse diamond



Protoplanetary nebulae  
The Carbon Ring Nebula (Kwoon, Su, and Hirvink 1998)

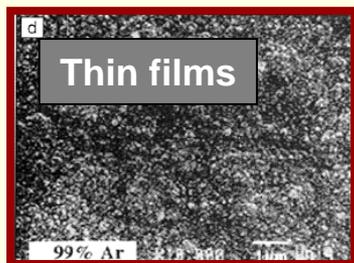
Extra-terrestrial diamond sources are often **ultra-disperse**, with grain sizes of  $\sim$  2-3 nm.

**Why ?**



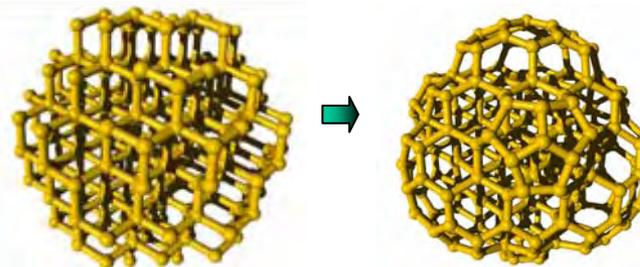
Meteorite fragments

Some diamond films and detonation residues may also be ultra-disperse.

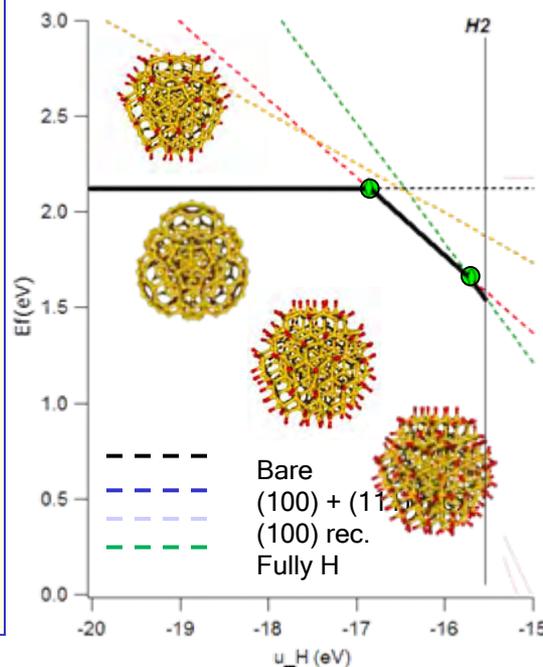


Thin films

**Detonation residues**



**Model for bare nanodiamonds** obtained by ab-initio MD (J.Y.Raty, G.Galli, A.Van Buuren and L.Terminello, PRL 03)

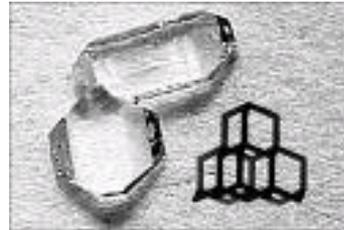
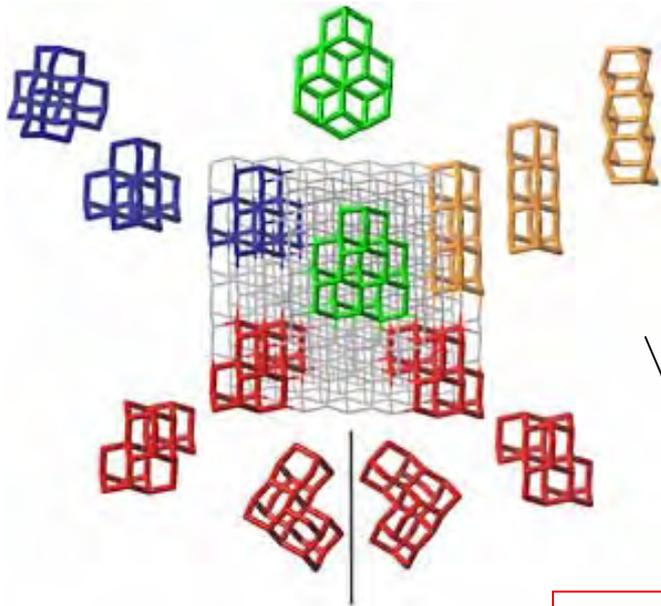


**Ab-initio** calculations of formation energies of nanodiamond as a function of hydrogen content explain **enhanced stability of bare nanodiamonds in 2-3 nm range**, and diamond ultradispersivity at the nanoscale.

(J.Y.Raty, G.Galli, Nature Materials 2003)



# Hydrogenated diamond clusters (diamondoids) found in oil



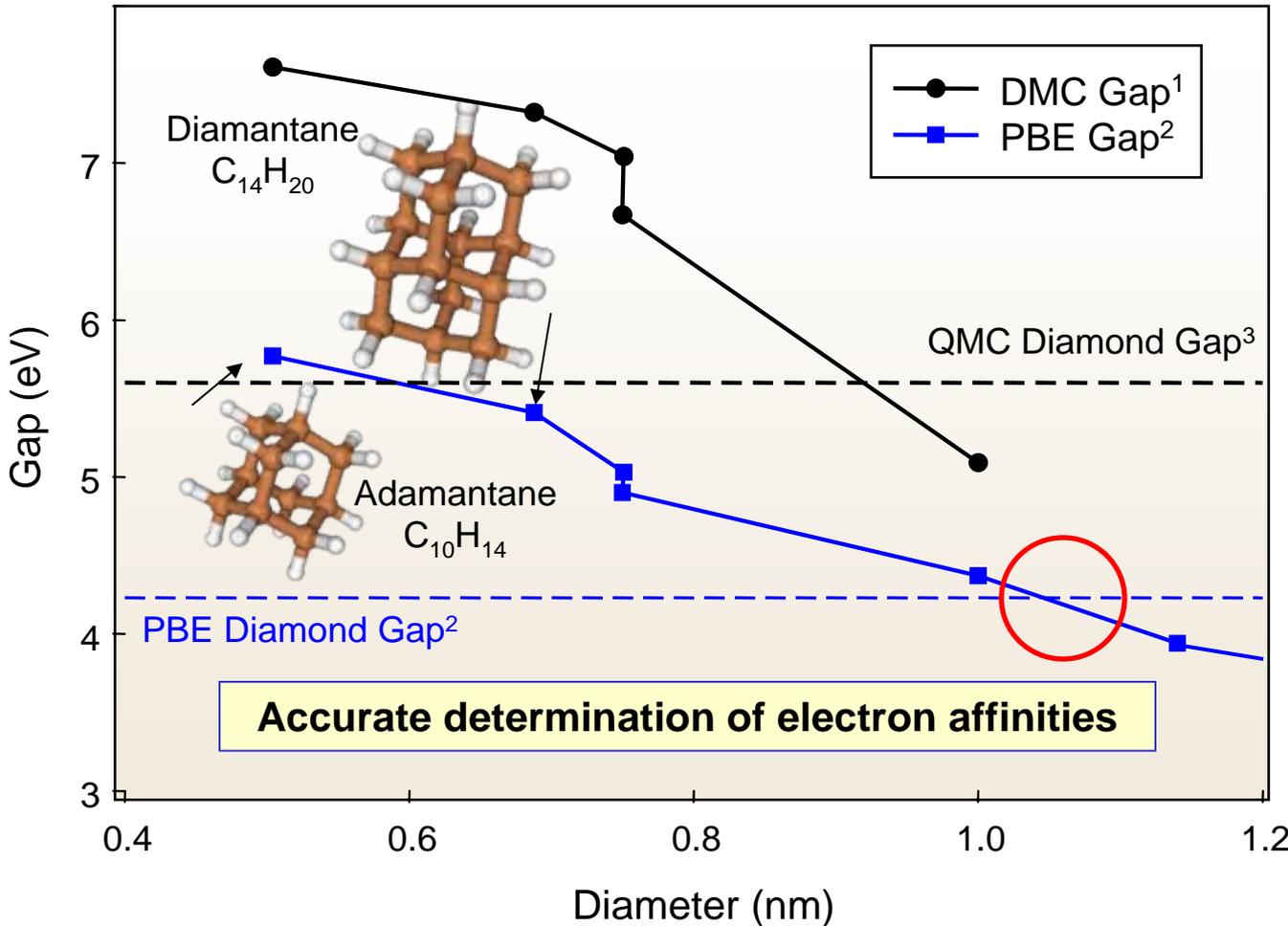
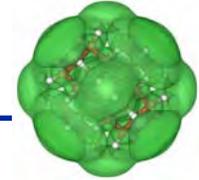
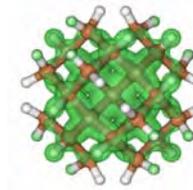
Functionalization (thiol groups) and deposition on Au and Ag surfaces

Chevron-Stanford project,  
2006-



STM images of tetramantane:  
M.Crommie et al. (UCB, COINS)

# Optical properties of diamondoids



**Prediction:**  
**small diamondoids have negative electron affinity**  
**→ electron emitters**

**Confirmed by experiments:**  
**Shen's group, Stanford, Science, 2007**

[1] N. Drummond, A. Williamson, R. Needs and G. Galli, PRL 2005

[2] J.-Y. Raty, G. Galli, C. Bostedt, T. W. van Buuren, and L.J. Terminello, Phys. Rev. Lett., 90, 037401 (2003)

[3] M. Towler, R.Q. Hood and R.J. Needs, Phys. Rev. B, 62, 2330 (2000)





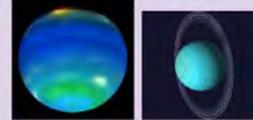
# High pressure phase diagram of carbon **predicted** by *ab-initio* molecular dynamics

- Melting at high pressure
- Change in electronic and bonding properties upon melting

## Astrophysics and Planetary Science

State of Carbon in:

- Giant planets

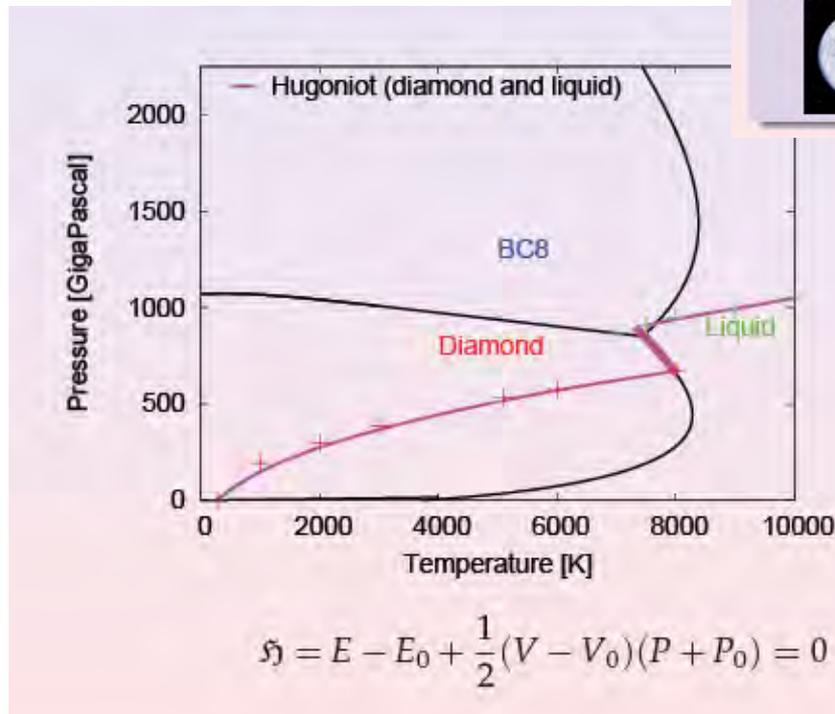
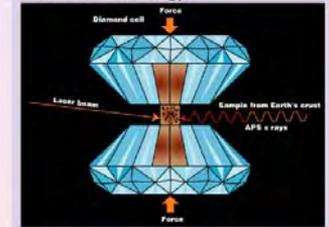


- White Dwarf stars



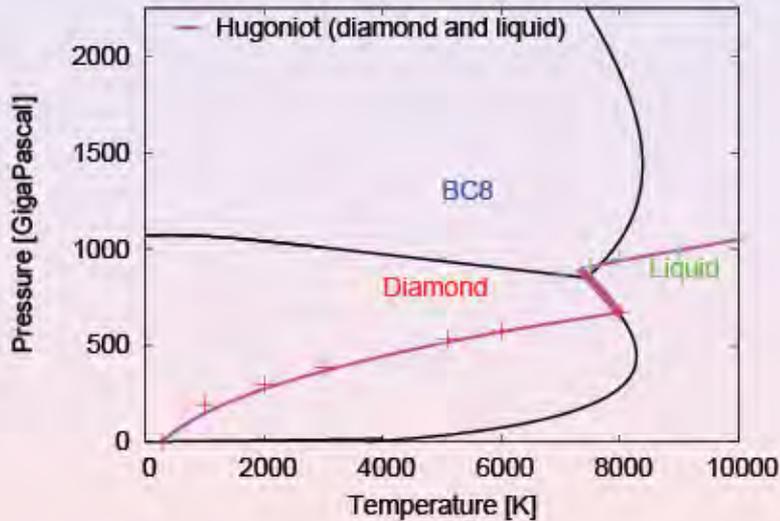
## Technological applications: High-pressure research

Theoretical limit of diamond based technology

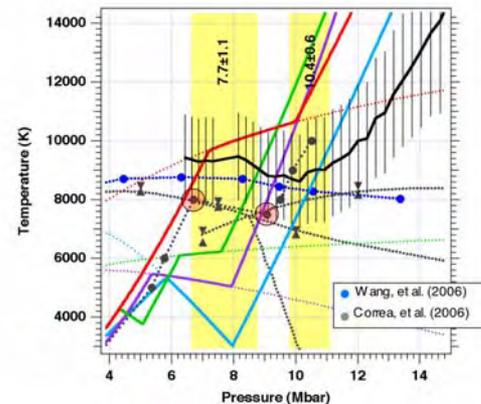
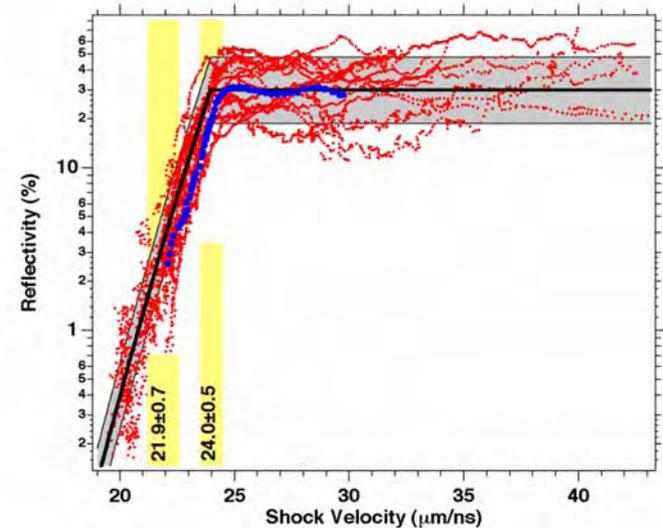




# Recent shock experiments **confirmed** slope of diamond melting line and metallization upon melting



$$\mathcal{H} = E - E_0 + \frac{1}{2}(V - V_0)(P + P_0) = 0$$

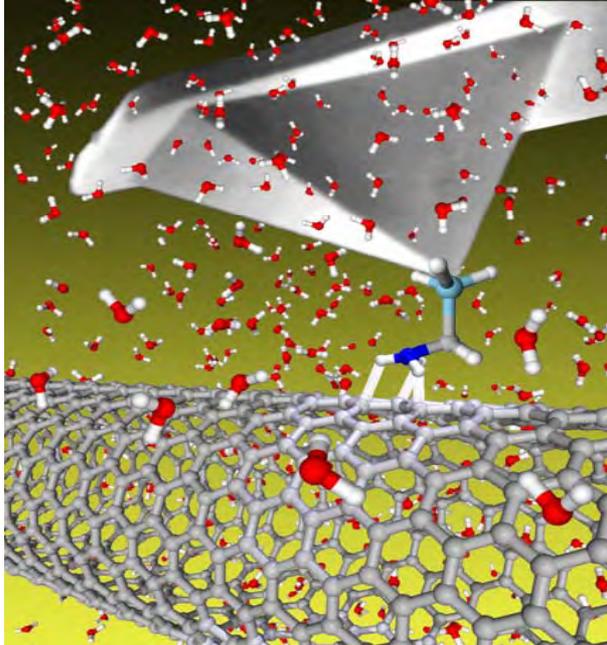


*Ab-initio* predictions used to interpret and guide current laser shock experiments

P.Loubyere et al.  
Nature Materials  
2007



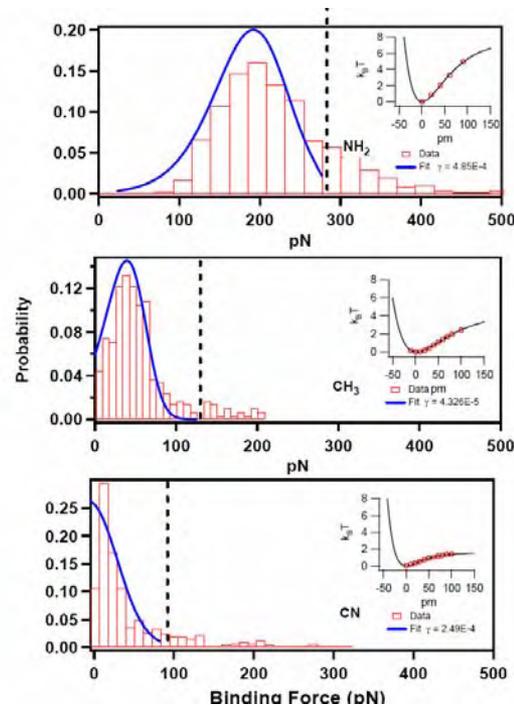
# Single molecule interaction with Carbon Nanotubes



Nature Nanotec. 2007

**Theory helped interpret and design experiments**

Solvent effects on AFM measurements



Comparison with experiment in the presence of solvent



# Outline

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- **Is computational materials science any good? Any promising progress?**

*Yes!*

—A couple of examples

– **Validation**

– **Impact**





# Outline

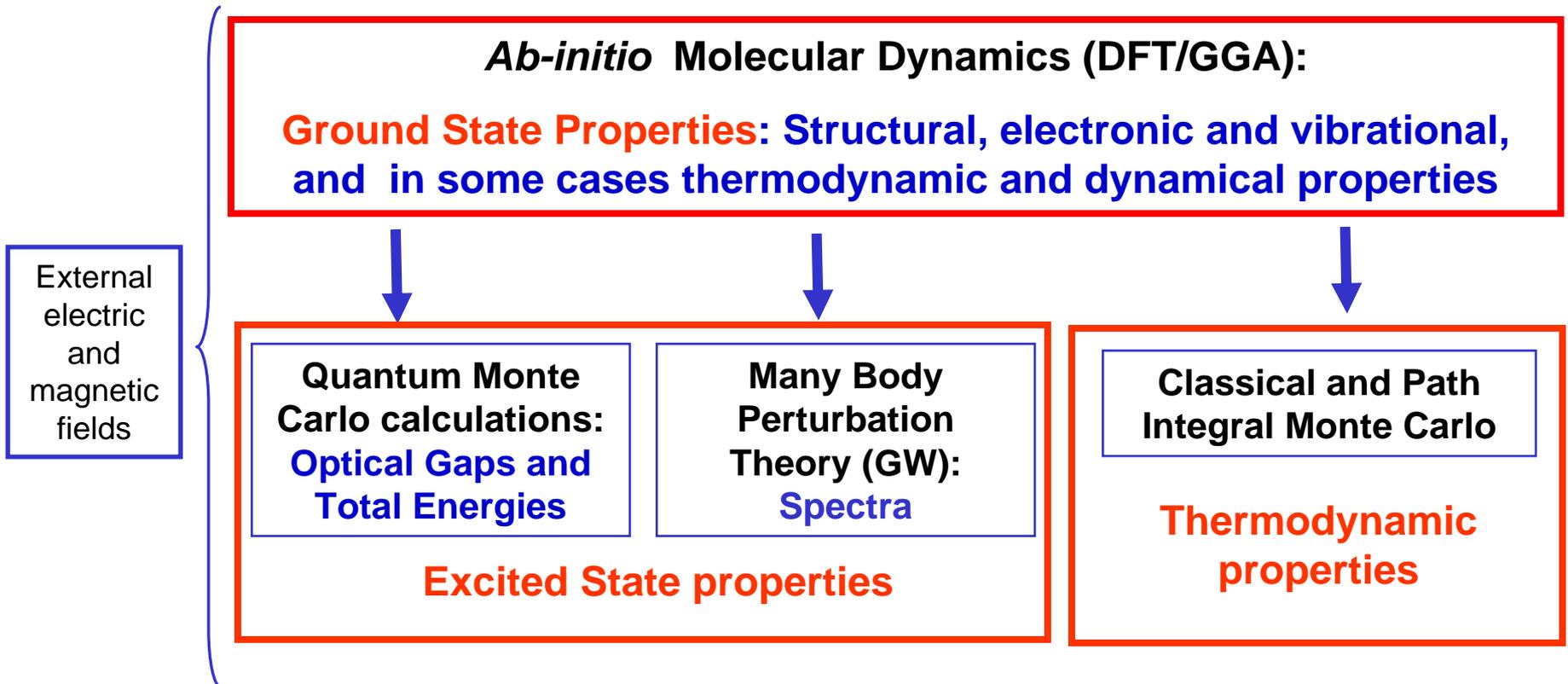
---

- Is computational materials science any good? Any promising progress?
  - A couple of examples
    - Validation
    - Impact
- Our strategy for predictive computations
  - On different fronts, in parallel
- Some grand challenge applications





# Predictive modeling with controllable accuracy



## *Ab-initio* Molecular Dynamics (DFT/GGA):



**Ground State Properties:** Structural, electronic and vibrational, and in some cases thermodynamic and dynamical properties

Solve set of  $N$  coupled, non linear partial differential equations self-consistently, using iterative algorithms, subject to orthonormality constraints.  $N = \#$  of electrons

$$\left\{ \begin{array}{l} -\Delta\varphi_i + V(\rho, \mathbf{r})\varphi_i = \varepsilon_i\varphi_i \quad i = 1 \dots N_{\text{el}} \\ V(\rho, \mathbf{r}) = V_{\text{ion}}(\mathbf{r}) + \int \frac{\rho(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + V_{\text{XC}}(\rho(\mathbf{r}), \nabla\rho(\mathbf{r})) \\ \rho(\mathbf{r}) = \sum_{i=1}^{N_{\text{el}}} |\varphi_i(\mathbf{r})|^2 \\ \int \varphi_i^*(\mathbf{r}) \varphi_j(\mathbf{r}) d\mathbf{r} = \delta_{ij} \end{array} \right.$$

Plane-wave basis sets and pseudopotentials

The cost of solving the Kohn-Sham equations is eventually dominated by orthogonalization ( $O(N^3)$ )

## *Ab-initio* Molecular Dynamics (DFT/GGA):



**Ground State Properties:** Structural, electronic and vibrational, and in some cases thermodynamic and dynamical properties

Optimization of  $10^6$  to  $10^9$  degrees of freedom for  $10^4$  to  $10^6$  time steps

$$\left\{ \begin{array}{l} -\Delta\varphi_i + V(\rho, \mathbf{r})\varphi_i = \varepsilon_i\varphi_i \quad i = 1 \dots N_{\text{el}} \\ V(\rho, \mathbf{r}) = V_{\text{ion}}(\mathbf{r}) + \int \frac{\rho(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + V_{\text{XC}}(\rho(\mathbf{r}), \nabla\rho(\mathbf{r})) \\ \rho(\mathbf{r}) = \sum_{i=1}^{N_{\text{el}}} |\varphi_i(\mathbf{r})|^2 \\ \int \varphi_i^*(\mathbf{r}) \varphi_j(\mathbf{r}) d\mathbf{r} = \delta_{ij} \end{array} \right.$$

Plane-wave basis sets and pseudopotentials

The cost of solving the Kohn-Sham equations is eventually dominated by orthogonalization ( $O(N^3)$ )



# Focus on optimized Kohn-Sham solvers, reduced scaling algorithms and data compression

Qbox

Petascale architectures

Optimization of  $10^6$  to  $10^9$  degrees of freedom for  $10^4$  to  $10^6$  time steps

$$\left\{ \begin{array}{l} -\Delta\varphi_i + V(\rho, \mathbf{r})\varphi_i = \varepsilon_i\varphi_i \quad i = 1 \dots N_{\text{el}} \\ V(\rho, \mathbf{r}) = V_{\text{ion}}(\mathbf{r}) + \int \frac{\rho(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + V_{\text{XC}}(\rho(\mathbf{r}), \nabla\rho(\mathbf{r})) \\ \rho(\mathbf{r}) = \sum_{i=1}^{N_{\text{el}}} |\varphi_i(\mathbf{r})|^2 \\ \int \varphi_i^*(\mathbf{r}) \varphi_j(\mathbf{r}) d\mathbf{r} = \delta_{ij} \end{array} \right.$$

(\*)  
Gordon-  
Bell Award  
2006,  
F.Gygi et  
al.

Optimized Solvers: Z.Bai and F.Gygi;  
Reduced scaling algorithms: F.Gygi and J.-  
L.Fattebert; Data compression: F.Gygi





# A Subspace Bisection Algorithm for Electronic Structure Data Compression

---

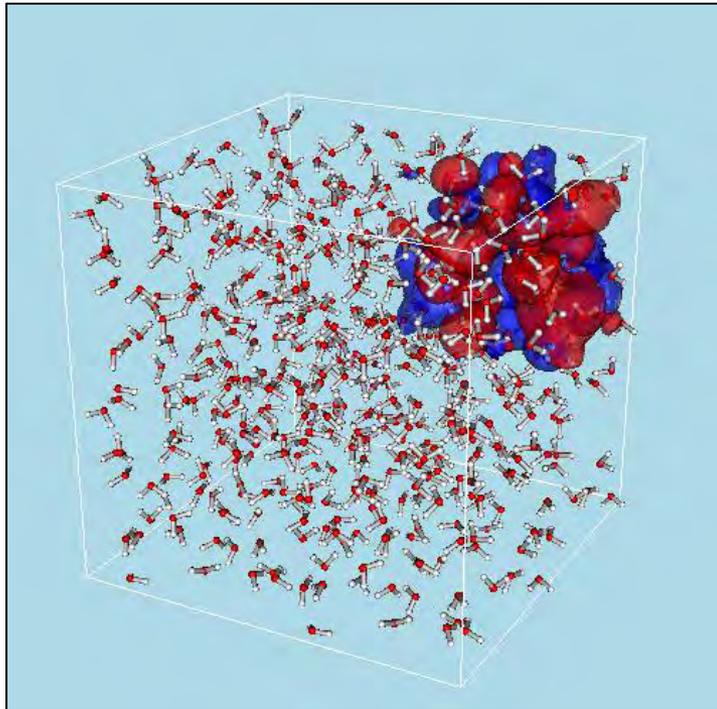
- The size of electronic structure data grows as  $O(N^2)$
- Goal: **Compressing electronic structure data while preserving accuracy**
- A compression algorithm based on the CS decomposition was tested on  $(H_2O)_{512}$  and a 304-atom CNT
- A compression ratio of **4.0 ( $(H_2O)_{512}$ ) and 2.7 (CNT)** can be achieved
- We are developing a recursive version of the algorithm to achieve higher compression ratios

The Subspace Bisection algorithm reduces I/O volume and restart file size on large-scale parallel platforms

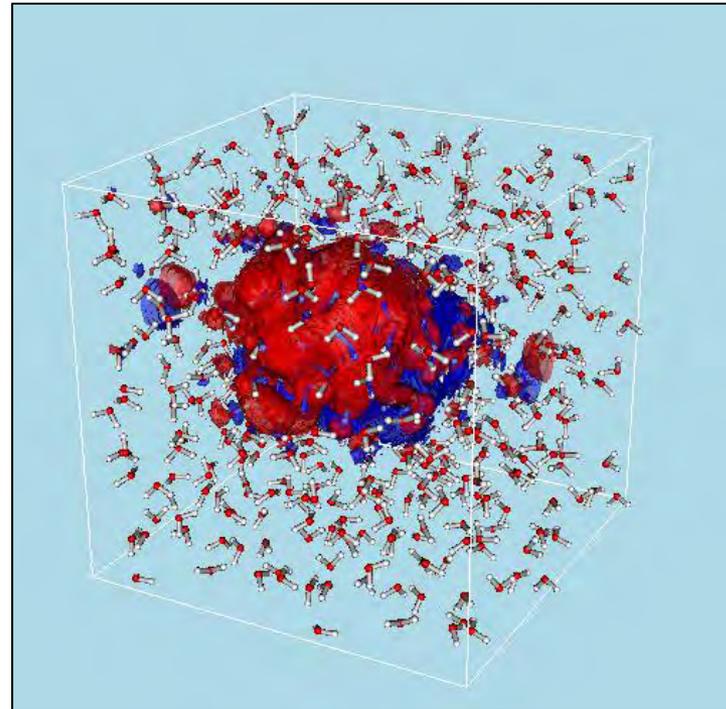


# (H<sub>2</sub>O)<sub>512</sub> states after bisection

localized

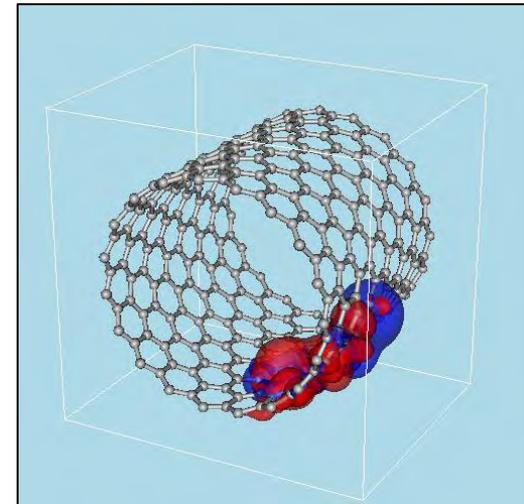
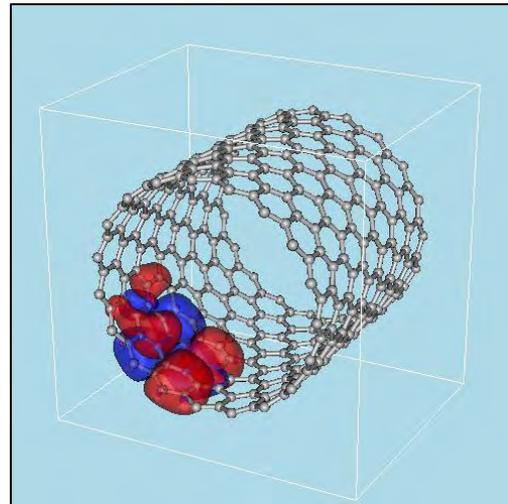


extended

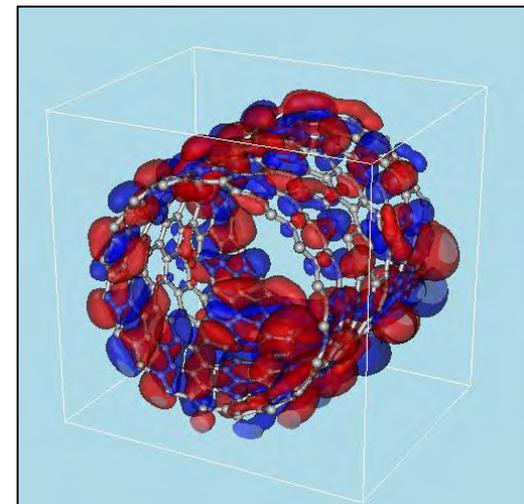
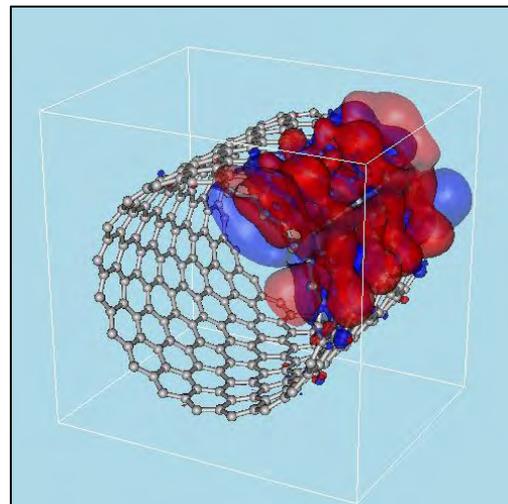


# CNT(19x0) states after bisection

localized

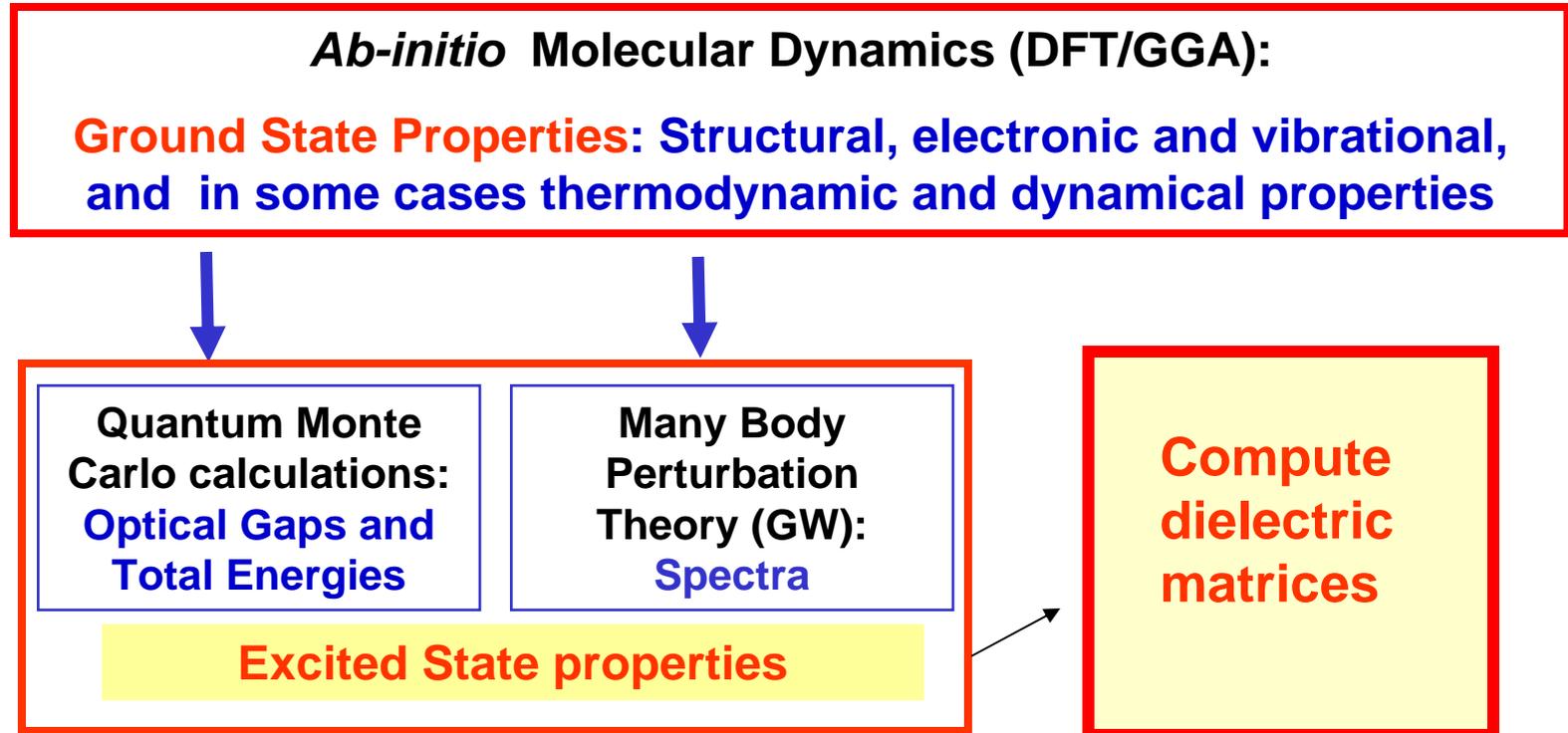


extended





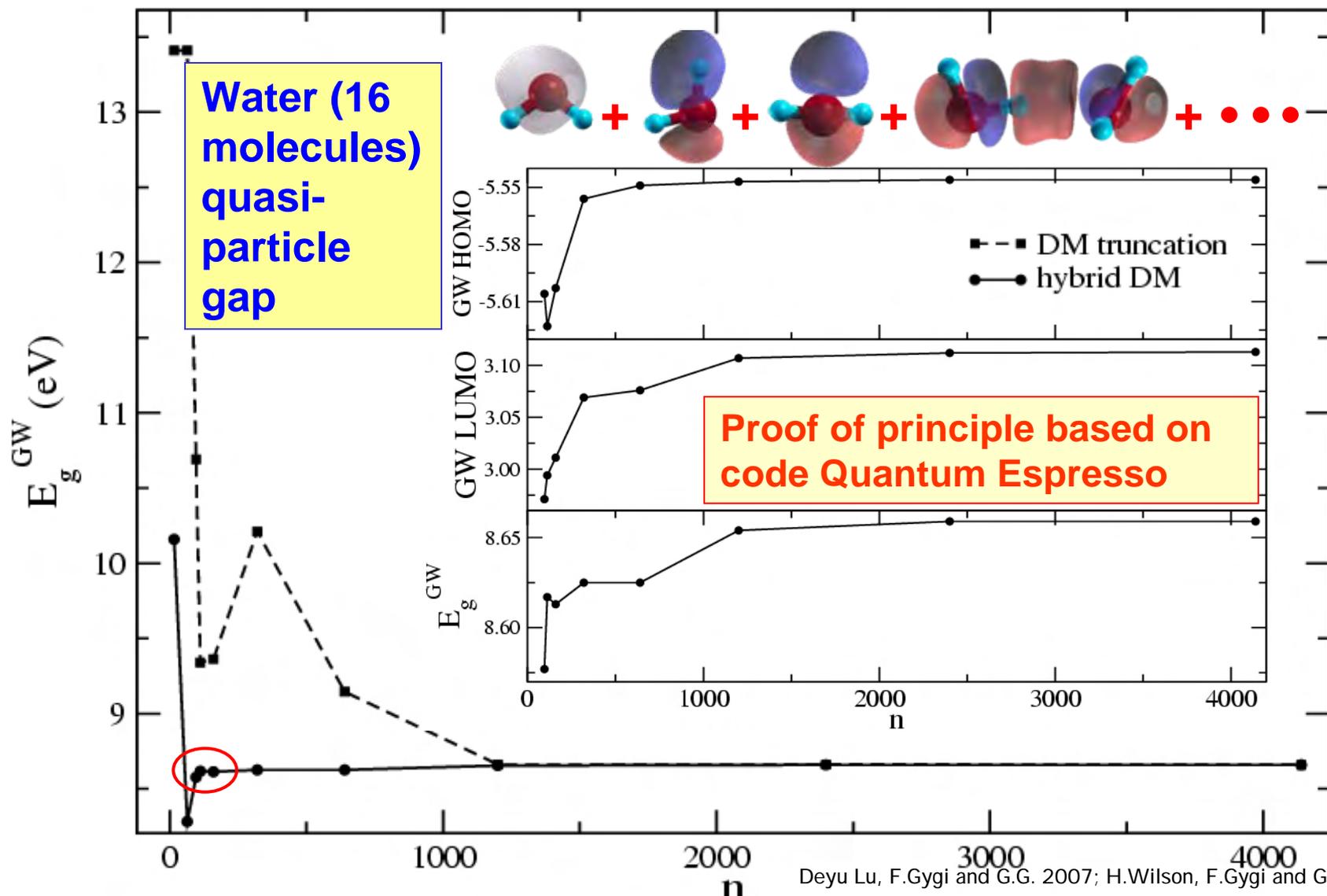
# Predictive modeling with controllable accuracy



$$\Sigma = iG(1,2)W(1^+,2)$$
$$W(1,2) = \int d(3) \epsilon^{-1}(1,3)v(3,2)$$

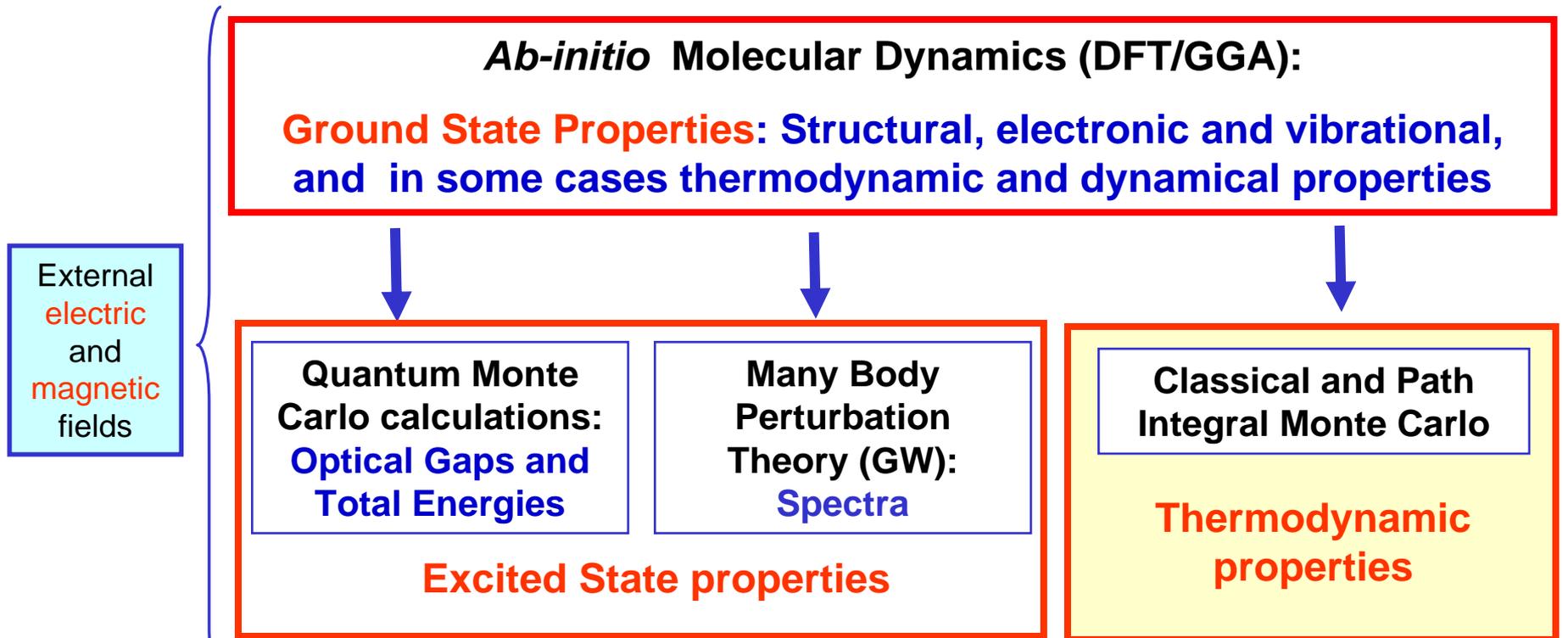


# GW energy gap from **approximate**, yet **accurate** and **non-empirical** dielectric matrices





# Predictive modeling with controllable accuracy



Generalization of ab-initio MD in a **magnetic field** to generic supercells: W. Cai, G.Galli et al. J.Comp.Phys.2007

**NMR spectra** calculations- N.Marzari et al. 2007

Ab-initio theory of metal-insulator interfaces in **finite electric field**, N. Spaldin et al. Phys. Rev. B (2007).

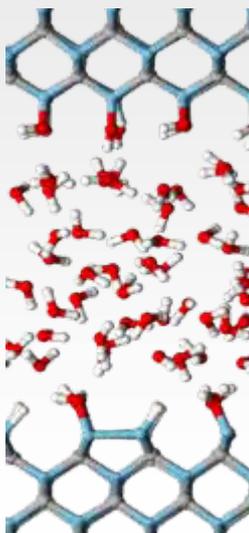
Coupling of codes Qbox and QMCPack in progress—proof of principle for classical MC achieved: D.Ceperley, F.Gygi et al.

# Water at the interface

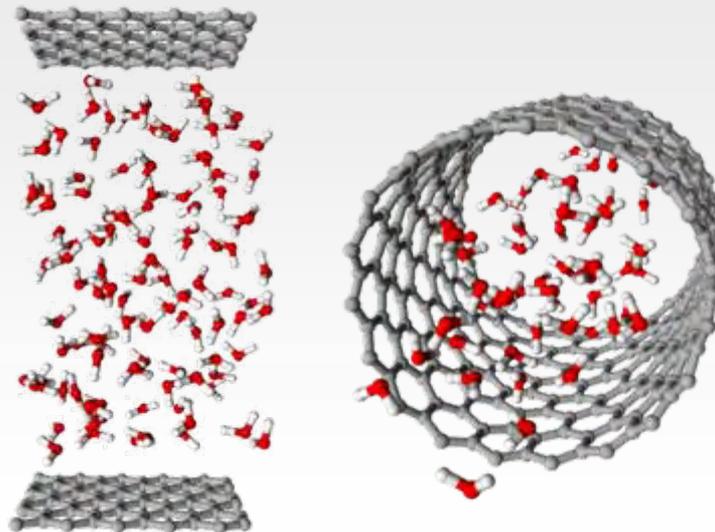


- We are studying water in different environments: from small ion and Si quantum dot solvation to wetting of surfaces in extended and confined media.
- We address fundamental issues regarding the atomic, electronic and dynamical properties of interfacial water.

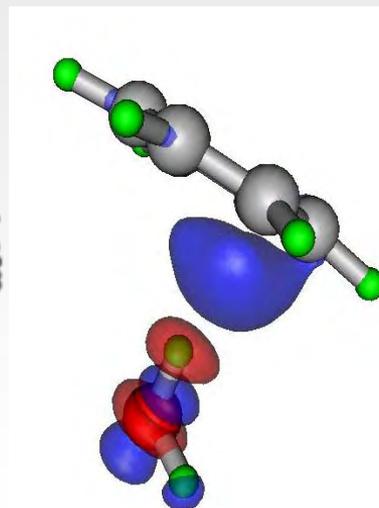
Hydrophilic surface SiC



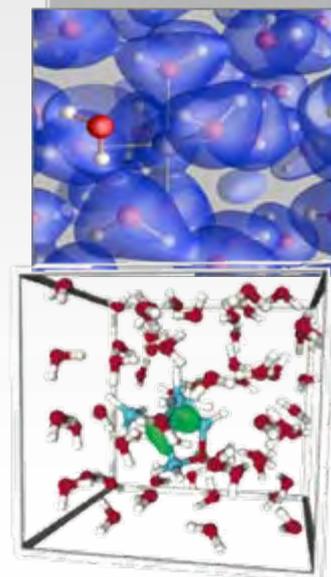
Hydrophobic surfaces – graphite, CNT and hydrogenated diamond



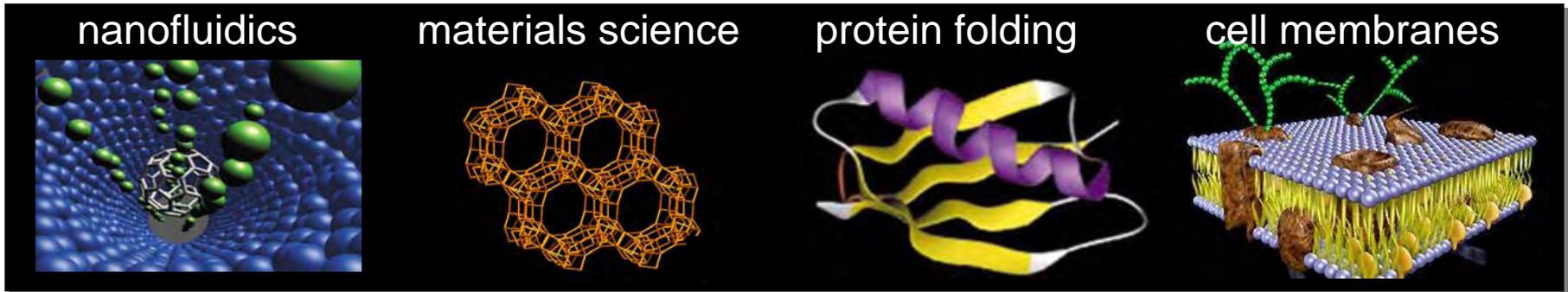
Hydrophobic solutes (benzene)



Solvation of ions and Si clusters



# Liquid water in confined media



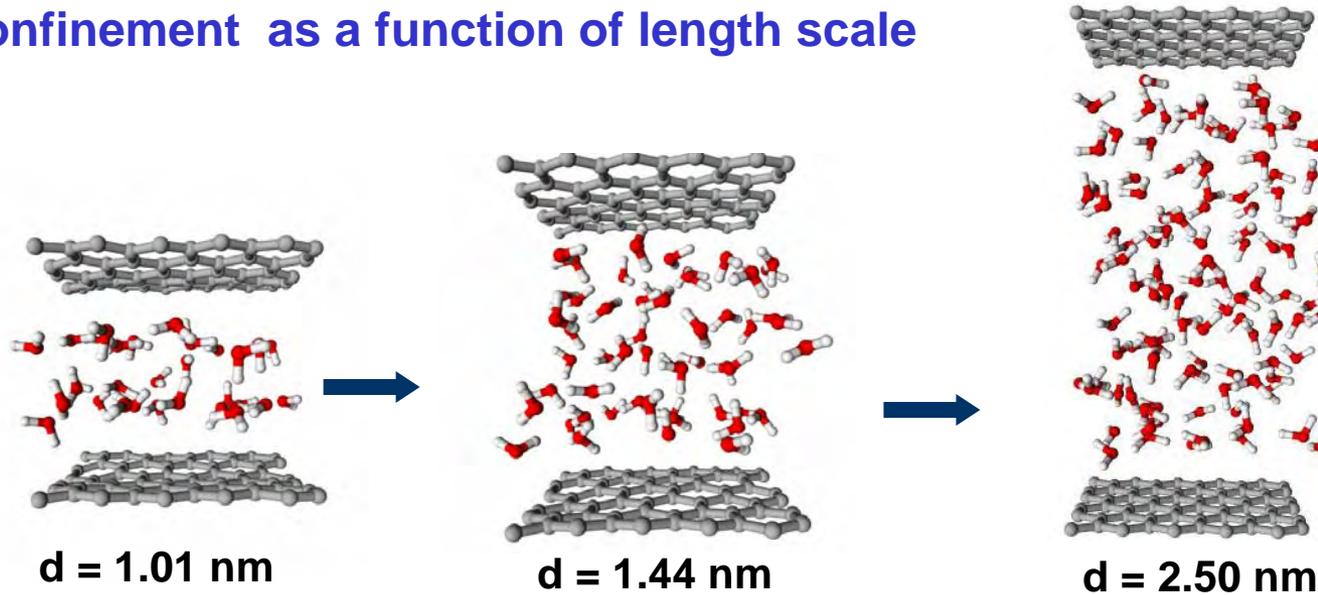
Understanding how the physical properties of bulk water (e.g. hydrogen bond network, diffusion, freezing point) are modified under confinement is relevant to a variety of outstanding scientific problems, including:

- » Studies of stability and enzymatic activity of proteins
- » Oil recovery
- » Nano-fluidics
- » Heterogeneous catalysis (role of water-substrate interaction)
- » Corrosion inhibition



# Computational Strategy

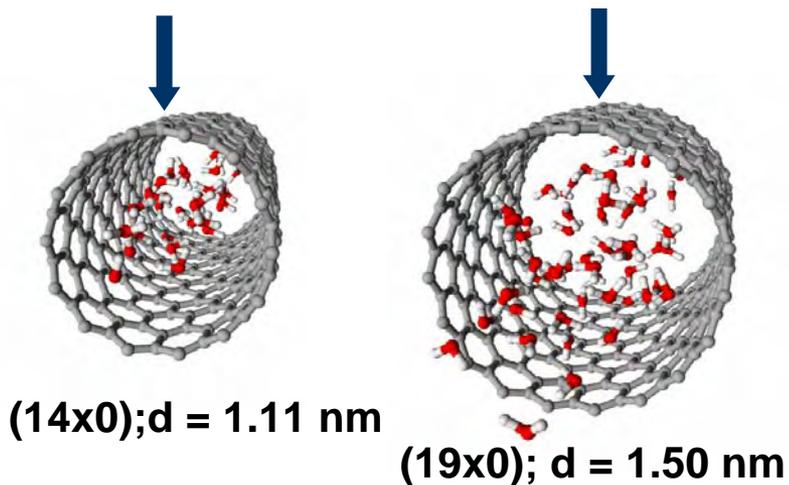
## Confinement as a function of length scale



•Simulation time :  
20-25 ps, except  
for graphene with  
 $d=2.50 \text{ nm}$  (10 ps).

•We used D  
instead of H for  
computational  
convenience.

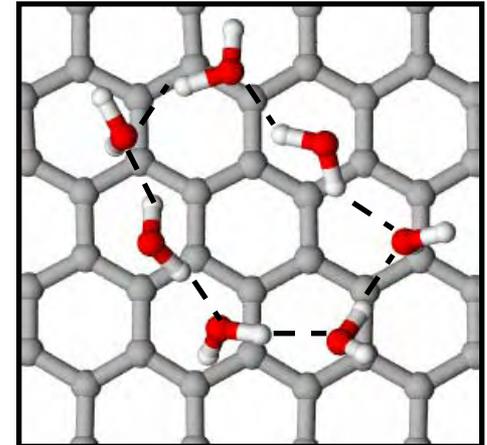
## Confinement in more than one dimension



Surface	# Water Molec.	# e <sup>-</sup>
None	64	512
SiC	57	1288
Graphite	32	496
Graphite	54 (49)	672
Graphite	108	1104
(14,0) CNT	34	1616
(19,0) CNT	54	1648

# Water confined within graphene and nanotubes

- **Perturbation induced by confinement is local**
- **No ice-like layer at the interface: liquid density increases**
- **Rarefaction and decrease of density away from the interface**
- **Dipole moment of water molecule at the interface decreases<sup>(\*)</sup> → lateral diffusion is enhanced and re-orientational dynamics is faster: Consistent with rapid flow in nanotubes detected in recent experiments**



**Sticky** yet fast water molecules at the interface

- **We predict that effects of OH bonds not engaged in hydrogen bonding and changes in librational modes are visible (although weak) in IR spectra.**

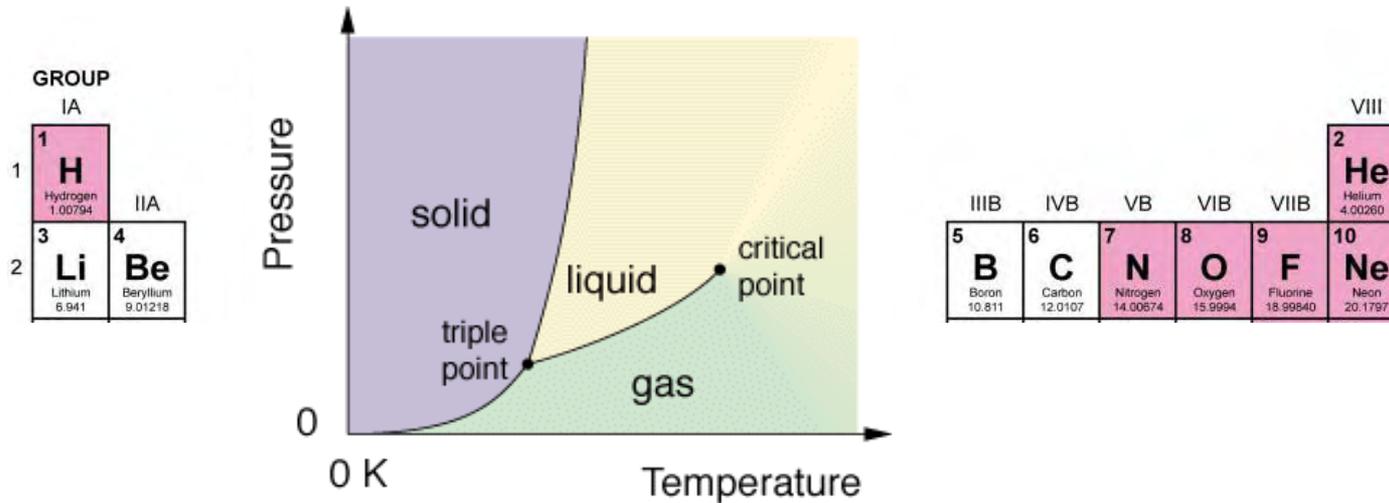
Complex **electronic interactions** occurs at the interface

G.Cicero, J.Grossman, E.Schwegler, F.Gygi and G.G. 2007 (submitted); M.Sharma, E.Schwegler, D.Donadio and G.G. 2007 (submitted).

(\*) Consistent with results obtained for Benzene and HFB in water: M.Allesch, E.Schwegler and G.G. JPC-B 2007.



# Understanding phase diagrams of elements and 'simple' compounds



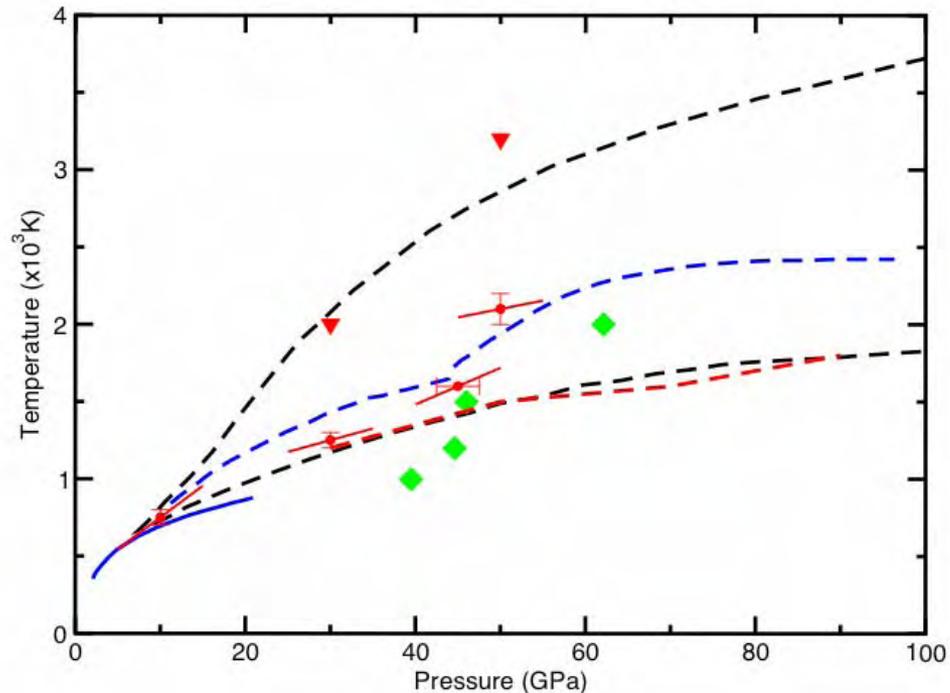
- Difficult to tame: Does diamond melt?
- Hydrogen: a liquid ground state at zero T and high P?
- A spin glass of oxygen molecules ?
- Boron: is there a crystalline ground state?





# Melting of water under pressure

- Interpretation of experiments and understanding of melting processes
- Solution of existing controversies



E.Schwegler, F.Gygi and G.G. 2007

Work on more complex oxides and metal-oxides by W.Pickett et al. - beyond DFT, using Dynamical Mean Field Theory



# Towards quantum engineering of materials



Simulations of **integrated materials** with **specific functions** for realistic devices



*Understand, predict and design*



Complexity of *design* lies both in **defining strategies and algorithms to compute novel properties** and in optimizing simultaneously several, **interrelated physical and chemical variables**.

Additional complexity **in software design and optimization**.





# Some of the most pressing scientific challenges of our age...

---

Researchers funded through the Office of Science are working on some of the most pressing scientific challenges of our age including: 1) Harnessing the power of microbial communities and plants for energy production from renewable sources, carbon sequestration, and environmental remediation; 2) Expanding the frontiers of nanotechnology to develop materials with unprecedented properties for widespread potential scientific, energy, and industrial applications; 3) Pursuing the breakthroughs in materials science, nanotechnology, biotechnology, and other fields needed to make solar energy more cost-effective; 4) Demonstrating the scientific and technological feasibility of creating and controlling a sustained burning plasma to generate energy, as the next step toward making fusion power a commercial reality; 5) Using advanced computation, simulation, and modeling to understand and predict the behavior of complex systems beyond the reach of some of our most powerful experimental probes, with potentially transformational impacts on a broad range of scientific and technological undertakings; 6) Understanding the origin of the universe and nature of dark matter and dark energy; and 7) Resolving key uncertainties and expanding the scientific foundation needed to understand, predict, and assess the potential effects of atmospheric carbon dioxide on climate and the environment.





# Many thanks to my collaborators

---

Z.Bai, F.Gygi and W.Pickett, UCD

W. Cai, Stanford U.

D.Ceperley, UIUC

N.Marzari, MIT

N.Spaldin, UCSB

E.Schwegler and J.-L.Fattebert, LLNL.

Thanks for your  
attention and for  
the kind invitation

**Computer time: *INCITE AWARD (ANL and IBM@Watson), LLNL, NERSC, SDSC***

